Biogeochemical Cycles Questions

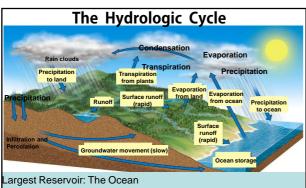
- Identify and describe the flow of nutrients in each biogeochemical cycle.
- Explain the impact that humans have on the biogeochemical cycles.
- > What happens to energy in an ecosystem?
- > What happens to matter in an ecosystem?

What are the major types of areas on earth? Atmosphere Geosphere Hydrosphere

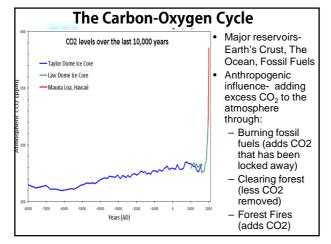
What Sustains Life on Earth? Abiotic chemicals (carbon dioxygen, nitrogen, minerals) Producers (plants) Producers (plants)

MATTER CYCLING IN ECOSYSTEMS

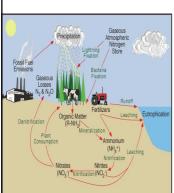
- > Nutrient Cycles: Global Recycling
 - Nutrients are the elements and compounds that organisms need to live, grow, and reproduce.
 - Nutritents: H20, CO2, N, P, S, O2 etc.
 - Biogeochemical cycles move these substances through air, water, soil, rock and living organisms.



- Anthropogenic influence-
 - Withdrawing large amounts of freshwater.
 - · Clearing vegetation and eroding soils.
 - Polluting surface and underground water..



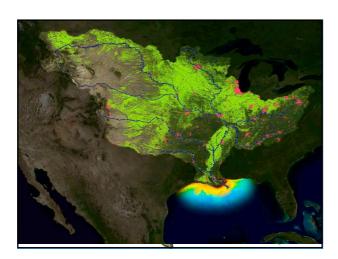
The Nitrogen Cycle

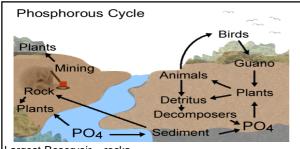


- Major Reservoir the air Anthropogenic influence-

- Creating synthetic nitrogen for fertilizer Contaminating ground water from nitrate in fertilizer.
- Excess nitrogen causing algal blooms and dead
- zones (eutrophication)
 Car/factory emissions
 creating acid rain.

Effects of Human Activities on the Nitrogen Cycle **Human activities** 18-24-6 such as production of fertilizers now fix more nitrogen than all natural sources combined.



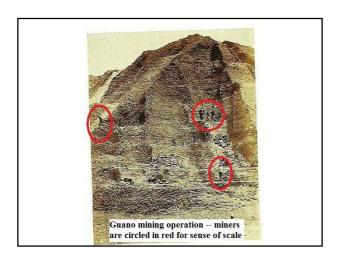


Largest Reservoir – rocks

No atmospheric phosphorous!

Anthropogenic influence- large amounts of phosphate from the earth to make fertilizer.

Excess fertilizer causes eutrophication (no oxygen in water) We reduce phosphorous in tropical soils by clearing forests.



The Sulfur Cycle					
Sulfur trioxide Sulfur dioxide Sulfur dioxide Sulfur dioxide Sulfur dioxide Sulfur dioxide Ammonium Sulfate Plants Dimethyl sulfide Ocean Metallic sulfide Sulfate salts Sulfate salts Hydrogen sulfide	Sulfur				

- Major Reservoir Earth's Crust
 Anthropogenic influence- Burning coal and oil, Refining petroleum, Convert sulfur-containing metallic ores into free metals such as copper, lead, and zinc all releases Sulfur Dioxide into the atmopshere.

 • Sulfur Dioxide mixes with water to make Sulfuric Acid Rain.

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